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## Application of the intramolecular aza-Wittig reaction to the synthesis of pyrido[2,3-d]pyrimidines

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Abstract—Pyrido[2,3-d]pyrimidines are synthesized in a two-step procedure from amides and tetrazolo[1,5-a]pyridine-8-carbonyl chloride. Reaction of the crude imides with triphenylphosphine effects an intramolecular aza-Wittig reaction to afford a variety of substituted pyrido[2,3-d]pyrimidines in good to moderate yields (30–76%).  $© 2006 Elsevier Ltd. All rights reserved.$ 

The beginnings of the aza-Wittig reaction are rooted in the work of Staudinger who synthesized the first iminophosphorane by reaction of tertiary phosphines and organic azides in the early  $1900s$  $1900s$ .<sup>1</sup> The reaction of iminophosphoranes with carbonyl compounds, particularly in an intramolecular sense, provides an effective approach to  $C=N$  bond formation. Consequently, the use of iminophosphoranes as a synthetic intermediate toward a variety of nitrogen containing heterocycles has become commonplace.<sup>[2](#page-2-0)</sup>

In the course of our investigations in the area of autoimmune diseases, $3$  we required an alternative route to synthesize compounds that contained a pyrido[2,3-d]pyrimidine core. Eguchi had previously reported that iminophosphoranes derived from 2-aminonicotinicamides react with a variety of acid chlorides to furnish pyrido[2,3-d]pyrimidines in excellent yields.<sup>[4](#page-2-0)</sup> The reaction was believed to proceed through an intermolecular aza-Wittig reaction followed by cyclization of the resultant imidoyl chloride 1 (Scheme 1). However, this method requires 200 mol % acyl chloride, and is limited to situations where the acyl chloride is stable or simple to prepare. Furthermore, this method is not applicable to substrates in which  $R^1$  and  $R^2$  are tethered. An alternative method that could overcome these drawbacks would be to perform the coupling between pyridine acyl chloride 2 and various amides, thus allowing access to



Scheme 1. Synthesis of pyrido[2,3-d]pyrimidines.

cyclic pyridopyrimidines.<sup>[5](#page-2-0)</sup> To this end, we were pleased to find that the intramolecular aza-Wittig reaction indeed furnished the desired compounds in good to moderate yields. Herein, we report the participation of tetrazolo $[1,5-a]$  pyridine-8-carbonyl chloride (4) in the synthesis of pyrido $[2,3-d]$ pyrimidines by way of an intramolecular aza-Wittig reaction.

The synthesis of tetrazolo<sup>[1,5-a]</sup>pyridine-8-carbonyl chloride (4) is summarized in [Scheme 2](#page-1-0). Beginning with

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Scheme 2. Synthesis of tetrazolo<sup>[1,5-a]pyridine-8-carbonyl chloride.</sup>

2-chloronicotinic acid, treatment with sodium azide afforded tetrazole 3 in 65% yield. Notably, the equilibrium in water between tetrazole 3 and its isomeric 2 azidopyridine lies exclusively at the tetrazole, consistent with the earlier characterization data reported for this compound.[6](#page-2-0) Reaction of 3 with oxalyl chloride afforded tetrazolo[1,5-a]pyridine-8-carbonyl chloride  $(4)$  in 95% isolated yield as a fine powder that is stable when stored at  $-20$  °C under argon (Scheme 2).<sup>[7](#page-2-0)</sup>

In a two-step process, amides were coupled with tetra $zolo[1,5-a]$  pyridine-8-carbonyl chloride (4) and cyclized

to generate pyrido $[2,3-d]$ pyrimidines in good to moderate yields  $(Table 1)$ .<sup>[8](#page-2-0)</sup> N-acylation of the amide was achieved through the use of  $Et_3N/DMAP$  which afforded the desired imides [\(Scheme 3\)](#page-2-0). Interestingly, the treatment of amides with NaH/THF or NaH/DMF followed by the addition of tetrazolo<sup>[1,5-a]pyridine-8-car-</sup> bonyl chloride (4) gave a mixture of  $N$ - and  $O$ -coupled products. The crude imide was used directly in the cyclization to generate the desired products (Table 1). Entries 1, 2, and 3 illustrate the potential of this method to synthesize cyclic pyridopyrimidines, a class of compounds that are particularly challenging to access using existing technologies.[9](#page-2-0) Electron-rich aromatic groups were well tolerated at  $R^1$ ; however, coupling of amides possessing electron-deficient aromatic groups at  $R<sup>1</sup>$  (p-iodophenyl, p-cyanophenyl) suffered from poor conversion primarily due to imide hydrolysis.[10](#page-2-0) Alkyl groups larger than methyl suffered from poor conversion. With respect to  $\mathbb{R}^2$ , aromatic, aliphatic groups and hydrogen were suitable substrates (entries 4, 5, and 6).

Table 1. Two-step approach to the synthesis of pyrido[2,3-d]pyrimidines





<sup>a</sup> Difficulty purifying from P(O)Ph<sub>3</sub> in hexanes:ethyl acetate chromatography. b Reaction conducted in THF at 60 °C.

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Scheme 3. Proposed mechanism to pyrido<sup>[2,3-d]</sup>pyrimidine formation.

Although no mechanistic studies have been conducted, one reasonable proposal for this sequence begins with the standard coupling to form imide 5a/5b. Treatment of 5a/5b with triphenylphosphine could trap-out the desired iminophosphorane 6 from the equilibrium mixture.<sup>11</sup> Cyclization of iminophosphorane 6 would then lead to the desired product.

In summary, a new approach to the synthesis of pyridopyrimidines has been developed. Utilizing tetrazolo $[1,5-a]$ pyridine-8-carbonyl chloride (4), imides were conveniently synthesized and transformed via an aza-Wittig reaction to furnish pyrido[2,3-d]pyrimidines in good to moderate yields. This two-step procedure offers an attractive alternative to conventional methods, and is particularly useful for the generation of cyclic pyridopyrimidines.

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- 7. Representative procedure: To a solution of 2-chloronicotinic acid  $(40.0 \text{ g}, 254 \text{ mmol})$  in DMSO  $(150 \text{ mL})$  was added sodium azide (16.5 g, 254 mmol) and the resulting mixture heated to 90  $\degree$ C for 3 h. The precipitated product was poured into 500 mL of acetone and filtered. The white filter cake was washed with an additional 1.5 L of acetone, and dried in a vacuum oven (room temperature) to obtain 3 (27.3 g, 65%) as a white solid. Synthesis of acyl chloride: To a slurry of the acid 3 (17.8 g, 108 mmol) in dichloromethane (150 mL) was added DMF (0.1 mL) followed by oxalyl chloride (18.9 mL, 217 mmol) slowly. The reaction mixture was stirred for 3 h, concentrated in vacuo, and washed  $2 \times$  with hexane (50–75 mL). The slurry was concentrated in vacuo to afford  $4(19.6 g, 98%)$  as a fine gray powder. \*Compounds 3 and 4 have not been tested for their propensity as explosives.
- 8. Synthesis of pyridopyrimidine representative procedure (entry 3): To a solution of azepan-2-one (226 mg, 2 mmol) in dichloromethane (5 mL) was added triethylamine (0.84 mL, 6 mmol), tetrazolo[1,5-a]pyridine-8-carbonyl chloride (4)  $(746 \text{ mg}, 4 \text{ mmol})$ , and DMAP  $(45 \text{ mg},$ 0.2 mmol) and stirred for 16 h. The reaction mixture was diluted with dichloromethane (125 mL) and washed with 1 M HCl (75 mL) and  $2 \times$  water (75 mL). The organic layer was dried (MgSO<sub>4</sub>), filtered and concentrated in vacuo to afford crude imide. The crude imide was transferred to a 15 mL sealed tube and taken up in toluene (5 mL). Triphenylphosphine (640 mg, 1.2 mmol) was added and the reaction mixture heated to  $110\degree C$  for 16 h. The solution was concentrated in vacuo and purified by silica gel chromatography (4:1 ethylacetate:hexanes  $\rightarrow 100\%$ ethyl acetate) to afford the desired pyridopyrimidine (330 mg, 76%).
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